

# Surface Analysis of Zinc Hydroxystannate-Coated Hydrated Fillers

Miklós Mohai<sup>1</sup>, András Tóth<sup>1</sup>, Peter R. Hornsby<sup>2</sup>, Paul A. Cusack<sup>3</sup> and Matthew Cross<sup>3</sup>

<sup>1</sup> Research Laboratory of Materials and Environmental Chemistry, Chemical Research Center, Hungarian Academy of Sciences, P.O. Box 17, H-1525 Budapest, Hungary

<sup>2</sup> Wolfson Centre for Materials Processing, Brunel University, Uxbridge, Middlesex, UB8 3PH, UK

<sup>3</sup> International Tin Research Institute Limited, Kingston Lane, Uxbridge, Middlesex, UB8 3PJ, UK



## Introduction

Zinc hydroxystannate ( $Zn[Sn(OH)_6]$ , ZHS)-coated fillers are novel flame retardant and smoke suppressant additives for polymeric materials. The application of ZHS coating to various hydrated inorganic fillers, in particular magnesium hydroxide or aluminium hydroxide, allows significant reduction to be made in the overall filler loading, with no loss in their flame retardant properties.<sup>1-3</sup> In order to help to rationalise this effect, in this work a surface characterisation of zinc hydroxystannate-coated magnesium hydroxide and aluminium hydroxide fillers was performed by XPS, including the determination of the surface coverage and the average layer thickness values of the ZHS coating.

## Sample preparation

ZHS-coated fillers were prepared according to the 'standard' route as follows. In a typical example, 1000 g of alumina trihydroxide (ATH, Alcan SF4, produced by Alcan Chemicals, UK) filler were slurried in 8 litres of an aqueous solution containing 103 g sodium hydroxystannate. One litre of an aqueous solution, containing 53 g zinc chloride, was added dropwise into the slurry, and the resulting mixture was stirred for 2 hours. The resulting solid product was separated from the solution by centrifugation, washed three times with distilled water and dried in air at 110 °C. The dried cake was crushed in a mortar with pestle to give 1110 g of a fine white powder (ZHS-coated ATH). ZHS-coated magnesium hydroxide (MH, Magnifin H5, produced by Martinswerke AG, Germany) powders were prepared using a similar method.

## Surface analysis

### X-ray Photoelectron Spectroscopy

Kratos XSAM 800 spectrometer  
Mg  $K\alpha_{1,2}$  excitation  
Fixed Analyser Transmission, 80 and 40 eV pass energy  
Referencing:  $CH_3$  type C1s (BE = 285 eV)  
C1s, O1s, Zn2p, Mg2p or Al2p

Data processing: Kratos Vision2000 data system

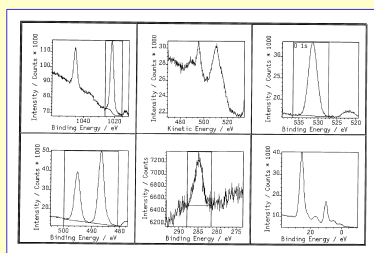
Model calculation: XPS MultiQuant, version 2.0<sup>4</sup>

Cross sections: Evans et al.<sup>5</sup>

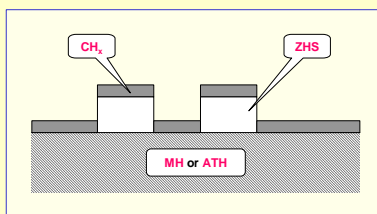
IMFP calculation: Seah and Dench<sup>6</sup>

Morphology: SEM<sup>1,2</sup>

## Characteristic spectra of zinc hydroxystannate



## The scheme of the Islands-on-Plane\* model applied for the quantification of the ZHS-coated MH and ATH samples



\* Lamelliform particles

## Calculation of concentration of ZHS

The theoretical ratio of the mass of the coating,  $m_c$  to the mass of the substrate,  $m_s$  is:

$$m_c / m_s = ahAr \quad (1)$$

where  $a$  - calculated coverage  
 $h$  - calculated layer thickness  
 $A$  - surface area of the substrate (6.2 m<sup>2</sup>/g by BET)  
 $r$  - mass density of the coating (3.3 g/cm<sup>3</sup> by product specification)

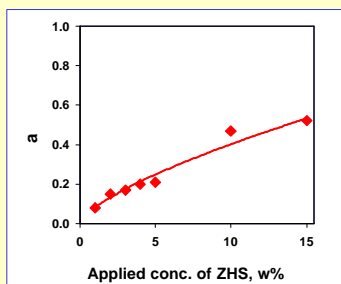
The bulk weight concentration ( $c$ ) of the coating is:

$$c = 100 m_c / (m_c + m_s) \quad (2)$$

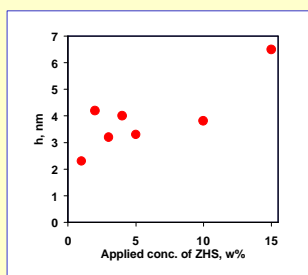
$$100 / c = (m_c + m_s) / m_c = 1 + m_s / m_c = 1 + 1 / AhAr \quad (3)$$

$$c = 100 / (1 + 1 / AhAr) \quad (4)$$

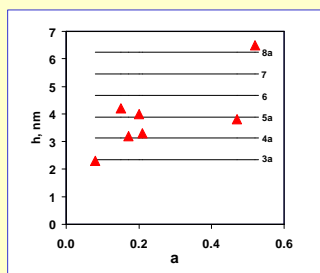
## Surface coverage ( $a$ ) of ATH by ZHS versus applied bulk concentration of ZHS



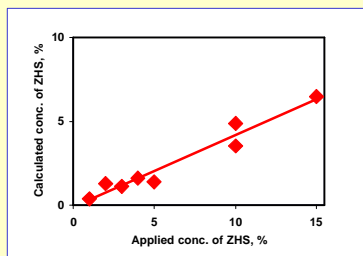
## Thickness ( $h$ ) of ZHS islands as a function of its applied bulk concentration



## Thickness ( $h$ ) of ZHS islands as a function surface coverage ( $a$ )



## Calculated versus applied concentrations of ZHS



## Summary

Magnesium hydroxide and aluminium hydroxide powder samples coated by zinc hydroxystannate (ZHS) to various extents have been studied by X-ray photoelectron spectroscopy. For quantification of the XPS results a model was applied, in which the substrate was covered by islands of ZHS, and this system was considered to be covered by a uniformly thin carbonaceous contamination overlayer. The surface coverage by ZHS and the thickness values of the carbonaceous and ZHS layers were determined for each sample by the recently developed XPS MultiQuant program. Relationships were established between surface coverage, layer thickness and applied bulk concentration of ZHS.

Low coating thickness values in the range of several nanometres have been obtained, which may account for the previously observed high flame retardant and smoke suppressant efficacy of zinc hydroxystannate-coated fillers when applied in various polymeric formulations.

## References

- Cusack PA, Patel B, Heer MS, Baggaley RG. *Intern. Patent Appl.* 1996; PCT/GB96/01475.
- Baggaley RG, Hornsby PR, Yahya R, Cusack PA, Monk AW. *Fire & Mater.* 1997; 21: 179.
- Cusack PA, Hornsby PR. *Journal of Vinyl & Additive Technology* 1999; 5: 21.
- Mohai M. *XPS MultiQuant for Windows User's Manual*. 1999-(2001); <http://www.chemres.hu/AKKL>
- Evans S, Pritchard RG, Thomas JM. *J. Electron Spectrosc. Relat. Phenom.* 1978; 14: 341.
- Seah MP, Dench WA. *Surf. Interface Anal.* 1979; 1: 2.

## Acknowledgement

This work was supported by the Commission of the European Communities (FP5 GROWTH Programme, Research Project "FLAMERET, New Surface Modified Flame Retarded Polymeric Systems to Improve Safety in Transportation and Other Areas", Contract Number G5RD-CT-1999-00120).