Surface Chemical Changes Induced by Low Energy Ion Bombardment in Chromium Nitride Layers



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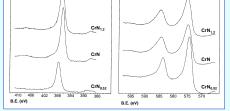


Introduction

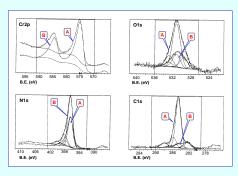
Chromium nitride CrN is an example of hard coatings providing high wear resistance combined with good tribological properties and excellent corrosion resistance. Therefore, it is one of the most universal tribological coating systems frequently used for forming and casting applications. Results on reactively sputtered ${\rm CrN}_{\rm x}$ coatings have been reported in many papers.

It is known that by varying the N₂ flow-rate in magnetron sputtering, coatings containing Cr, Cr(N) solid solution, Cr₂N and CrN phases can be deposited. Determination of the crystallographic phases present in these films is usually done by X-ray diffraction, which allows the identification of the overall crystalline phases. However, in the case of dual-phase film structures or very low grain sizes the exact determination of the individual phases by XRD can be ambiguous due to overlapping peaks or peak broadening. Thus, the aim of this work is to evaluate the composition and the chemical state of the elements in CrN_x films, deposited by magnetron sputtering at varying $N_2/(N_2+Ar),$ using X-ray photoelectron spectroscopy. Implications of the effect of ion bombardment on the composition and structure of the CrN films have been clearly demonstrated recently. This oriented our attention towards investigations of the effect of Ar* and N_2^{\star} bombardment on the CrN_ layers of different composition and structure. Conclusions were made upon the possible alterations of the Cr2N and CrN stoichiometry by these bombarding ions.





spectra of the CrN_{1+y} sample: Cr2p, O1s, N1s and C1s regions with the synthetic-line components of two states,





Aims

In this work three types of CrN_x layers, one close to Cr_2N , the other to CrN composition and a third with high nitrogen content CrN_{1+x^*} were prepared by reactive magnetron sputtering. The samples were subjected to bombardment by Ar^ and N_2^ ions applied in sequence. The relative atomic concentration and the chemical states of the elements in the surface layer were determined by X-ray photoelectron spectroscopy.

Coating	N ₂ /(Ar+N ₂)	E _i [eV]	Jį/Ja	t [µm]	Phases	d [nm]
CrN0.52	0.20	87	0.41	3.0	Cr ₂ N	16
CrN _{0.98}	0.31	32	0.61	3.3	CrN	60
CrN _{1.00}	0.53	32	0.84	3.2	CrN	18
CrN _{1+x}	0.71	32	1.32	3.2	CrN	10
CrN _{1+y}	0.83	32	1.32	5.5	CrN	9

Experimental

Film Deposition

Substrates: silicon (100), molybdenum, austenitic stainless steel Unbalanced Dro Magnetron CT target (193-99) & 14 Ora, Ar + N₂ (99.999 %) substrate temperature 300 °C. Base pressure 5 8 × 10⁴ Pa, working pressure, 0.4 Pa. Sputtering power density 3 W cm² The plasma parameters measured using a Hiden ESP Langmuir wire probe.

Sample Analysis

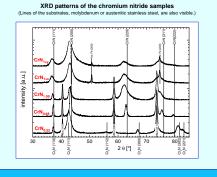
Film thickness: spherical abrasion test Chemical composition: Microspec WDX-3PC wavelength-dispersive electron-probe microanalysis (EPMA) XRD: Siemens D500 (Bragg-Brentano mode), Cu Ka, Pseudo-Voigt profile function

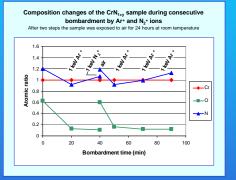
Ion Bombardment

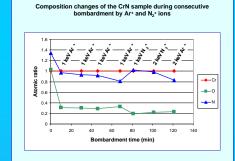
Kratos MacroBeam and Technorg Linda ion guns 0.5-5 keV, doses up to $10^{18}\,ions/cm^2$ 99.999 % N_2 and Ar

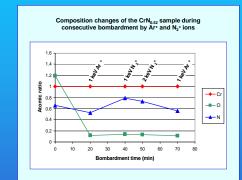
XPS Measurements

Kratos XSAM 800 spectrometer Mg Ka radiation, base pressure 10⁻¹⁰ mbar Fixed analyser transmission mode with 40 eV pass energy Shirley type background subtraction Kratos Vision 2000 and XPS MultiQuant softwares









Conclusions

- All chromium nitride samples were crystalline. The phase and chemical composition were in good agreement. The XRD lines of the super-stoichiometric samples showed a distinct peak broadening.
 - Ar⁺ ion bombardment could not reduce the nitrogen content of the $CrN_{0.52}$ sample while N_2^+ bombardment increased it up to CrN_{0.8} but complete conversion to CrN could not be achieved.
- Ar⁺ ion bombardment reduced the nitrogen content of the CrN sample to CrN_{0.8} and N₂⁺ bombardment increased it slightly above 1:1.
- Super-stoichiometric chromium nitride samples up to $\text{CrN}_{1,3}$ composition could be prepared by the applied method. The excess nitrogen could not be removed by Ar+ ion bombardment