

SURFACE MODIFICATION OF TRIPLEX PVD COATINGS FOR CUTTING TOOLS

A.I. Kovalev, D.L. Wainstein. Surface Phenomena Researches Group, CNIICHERMET, Moscow, Russia

E-mail: sprg@mail.sitek.ru G.S. Fox-Rabinovich. Department of Material Science and Engineering, McMaster University, Hamilton, Canada.

E-mail: gfox@mcmail.mcmaster.ca.

hanges in chemical composition and fine (at atomic level) structure characteristics of surface during lifetime of cutting tools were investigated. High-speed steel of M2 type was hardened by well-known 'duplex' technology (ion nitriding with TiN PVD coating) in the present wor to top layer of the duplex coating was modified by perfluoropolyether Z-DOL. There was found, that the process of the self-organizing is taking place during cutting on the surface of the hardened material. Several parallel processes are developing during the self-organizing. The e as follows: a) dissociation of chemical compounds of the multi-layered coating (TiN and perfluoropolyether) and intersive diffusion of nitrogen into the chip; b) forming of the stable amorphous-like protective layers (secondary structures, of Ti-O and Ti-F types) on the to rface. For tools with multi-layered coating, containing top antificitional Z-DOL layer, the surface damage is minimal in the running-in stage of the weart her resistance for the tool with studied multi-layered coating. MATERIALS AND METHODS

MA LEKIALS AND MELINUS this work we have studied the 'triplex' coating including ion nitriding of HSS base, nitride PVD coating deposition and top layer of PPFE libriorian. The hard coatings are relatively brittle ones and susceptible to intensive surface damage during the running-in stage of cutting too ear. Eventually, the main part of the hard coating already destroys at the initial phase of the wear prior to the begining of the stable (normal) stage of wear when the wear rate is lower by an order of magnitude compare to the running-in stage. This results in critical decrease of the ating wear resistance. That is why in this work the special antifrictional (adaptable) top layer with protects the hard coating during the initial (damaging) stage of wear. This is one of the most important goals for the wear resistant coatings. PFPE's are currently being used : bricants in industrial applications. These fluyids are attractive us lubricants because of their high thermal stability, low vapor pressures, and chemocal incremess. The perfourpoolyteher was applied as antifrictional material to deposit as a top layer on the surface of the dard coating reliquid lubricant is a 0.5 % solution of perflouropolyether Z-DOL in freon 113. PFPE has the following chemical structure: HO-CH₂-CF₂O-(CF₂O)-(CF₂O)-(CF₂O)-(D₂-CH₂-CH₁-CH₁. The PFPEF film was deposited by dipping into the 'hot' (at the temperature of boiling) solution. Triplex coating and its wear resistance characteristics





IMS spectra of TIN coating with Z-DOL anti-frictional top layer.)- spectrum of positive ions (before service); (b)- spectrum of gative ions(before service); (c)- spectrum of positive ions uting during 200 s).)- spec gative



800 700 600 500 400 300 E.e. Z-DOL anti ce; (2) 100 s; (3) 300 s; (4) 500 s; (5)



per independence of cuining inter (Sints dual), at 15 (0 + 90%) c + 160% (1), are anti-frictional PFPE layer prevents a damage of TiN hard atting and makes oxidation of the surface more gradual. In this se oxidisting of TiN is observed at 300-500 s of cutting, xidation of titanium nitride in our case is a beneficial process. tanium-based oxygen containing compound acts as a shield that otects the surface. That is why the emergence of increased antity of oxygen containing compound at transition from the ork-in stage to that of normal wear testifies to the self-organiza-n of "tool-workpicce" tribosystem. As a result, the wear tensity is reduced and this process enters its steady-state stage, rhber on, the quantity of oxide component remains practically uchanged as the cutting time grows (SIMS spectra (b), (c)).

On SIMS spectra the PFPE Z-Dol gives the following peaks: O⁻(16); F⁻(19); CF₂⁻²(25); CF₂O⁻²(33); F₂⁻(38 amu.). These results are testified that the PFPE lubricant is chemical neutral at initial stage of cutting. The composition of the coating is altered during cutting. The small intensity peaks at 64 and 86 amu are appeared after cutting during of 200s. Obviously, we can attribute these peaks with throbe decomposition of PFPE and TiN and forming of tianium oxide and TiF₂.

Element composition analysis on surface of the crater on the rake surface of the tool carried by means of AES shows that at initial stages of wear in the running-in zone three is a gradual dissociation of chemical compound of TiN-type under influence of considerable local stresses and temperatures generated on the surface of the coating during cutting. This process is observed in TiN coatings with and without using of PFPE top antifrictional layer. This manifests itself in gradual reduction of nitrogen concentration on the surface at the work-in stage. Simultaneously, the oxygen in the contact zone replaces the nitrogen. These effects

concentration on the surface at the work-in stage. Simultaneously, the oxygen in the contact zone replaces the nitrogen. These effects are clearly presented by a series of Auger-spectra (1-5), obtained from the surface of wear crater at different time of cutting. Significant increase in intensity of Auger-Ines for iron at con-siderable time of cutting is brought about by the adhesion of steel on the surface of cutting tool and by the destruction of coating at the final stage of cutting. There is a good correlation between the change in wear resistance of cutting tool and the element composition of the coating. Depletion of coating surface by N and its enrichment by O during the transition from the running-in stage to steady stage of normal wear is connected with the dissociation of Ti-N, adsorption of O from the environment, and development of Ti-O-like compound. The oxidizing of Thi is observed at initial stage of cutting in ordinary TiN coating (SIMS spectra).

e of PEPE molecule (frontal and lateral view)





(a) HREELS spectrum of TiN coating on HSS. (b) HREELS spectrum of PFPE Z-DOL on TiN coating. In both cases the primary energy was 7.0 eV



ant of (a); nant of EELFAS data below KLL line of N for the

The physical mechanism of triplex coating wear

The physical mechanism of triplex coating wear We can suggest the following mechanism of the hard coating wear with the top PFPE layer. The polymer molecula contacts to the surface of the TiN coating by positive polarized thyse polarized theorem in CF groups push back out of electron cloud surrounding the cutting tool/workpiese interface. The similar process of generation of surface potential under PFPE lubrication was shown elsewhere [9]. Due to this electrico-statical effect PFPE lubrican significantly decrease the friction parameter at the cutting tool/workpiese interface. The accumulation of structural imperfections in the tool surface is decreased. There has been established that lubricant degradeted at initial stage of cutting. Triboemission of the negatively and positively charged particles and surface potential due to tribocharging and tribodecomposition of PFPE are also shown earlier [9]. This process results in ionic fluorine adsorption and metal-fluoride interaction. After 200 s of cutting TFP; is appeared. This process is accomponied with oxidizing of TiN and mass transfere of hitrogen to the chip amorphous-like structure. Oxide and fluoride of titanium are posseess high antifrictional properties of coated cutting tool, how merous ways. In this case friction control for engineered cutting, than ying an amorphous-like structure. Oxide and fluoride of titanium are posseess high antifriction appretise of coated cutting tool. These processes are the basis of self-organizing under cutting. Hard TiN coating with self-organizing process is associated with the running-in stage of wear. During this stage of self-organization, the wear process gradually stabilizes and finally transforms to stable (or normal) stage. It is very important to prevent surface damage and promote self-organization on the surface damage of the wear process gradually stabilizes and finally transforms to stable (or normal) stage. It is very important to prevent surface damage and promote self-organization on the surface damage of the w

Conclusions

Conclusions The reserches made it possible to represent the change in the structure of complex engineered coating applied to the HSS. Cutting tools steel of M2 type hardened by 'duplex' technology (ion nitriding with ic plasma TN coating and top layer of perfluoropolyether Z-DOL) were used in the present work. At the initial stage of cutting the PFPE lubricant decreases the friction coefficient and significantly decrease the structural damages on the tool surface. We can guess that this effect is a result of electro-static screening by the lubricant at the cutting tool/workpicee interface that impoves ability of the surface to dissipa energy during friction. Titanium nitride of the hard PVD coating is oxydizing during cutting. The tribodecomposition of Z-DOL is the reason of forming the titanium-based fluoride and oxygen containir compounds on the tool surface. Forming at friction of amorphous-like compounds leads to low friction and heat transfere coefficients of the coating. Formation of the thin films of these compounds improves th thermal and friction conditions at the interface and stabilizes the friction process. Dissocoation of TNC coating is accompanied with vigorous diffusion of nitrogen into chips.

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HREELS of PFPE

OI FFFL Figures shows the vibrational spectrum of TiN coating without (a with Z-DOL lubricant deposited on tool surface. The surface cleaned prior to examination by heating in high vacuum X10⁴⁴ at 473 K for 15 min. Vibrational energy loss peaks for PVD coating deposited on HSS are observed at 10 and 370 meV. Th assimetrical peak (110 meV) is observed at characteristic energy the transverse optical phonons with out-of-plane displacements in the spectra. In addition ,we detected a broad loss peak central 185 meV due to optical phonons with in-plane displacement feature in high energy loss portion of the spectrum at 370 m attributed to an overtone of the loss at 185 meV.

attributed to an overtone of the loss at 185 meV. Vibrational energy loss peaks for Z-DOL deposited on TiN are observed at 100, 159 meV (Sce Fig. b). In additional intensity los spectra was detected at the energies in the range of 350 – 1400 m The intence loss at 159 meV is tipical for perfluoropolyethers and associated with the coupled stretching motion of the (CF₂–O-CF₂ linkage (7,8]. The peaks of losses at 100 meV and 350 meV are analogous to 110 meV and 370 meV ones on the TiN spectrum. T softened phonons come from the weakening Ti–N bonds of substi owing to the additional chemical bonds of N with H of the (CH₂– groups of PFPE film. Periodical fine structure of the loss spectra i energy range 350–1400 meV is probably overtone phonons spect stipulated by collective surface vibrations and explained by high desordered structure of PFPE film.





the FELES Fourier transformants of the EELFS below wear hollow: a - 30 s; b - 180 s; c - 2100 s.[1]